

# PERIODIC TABLE Atomic Properties of the Elements

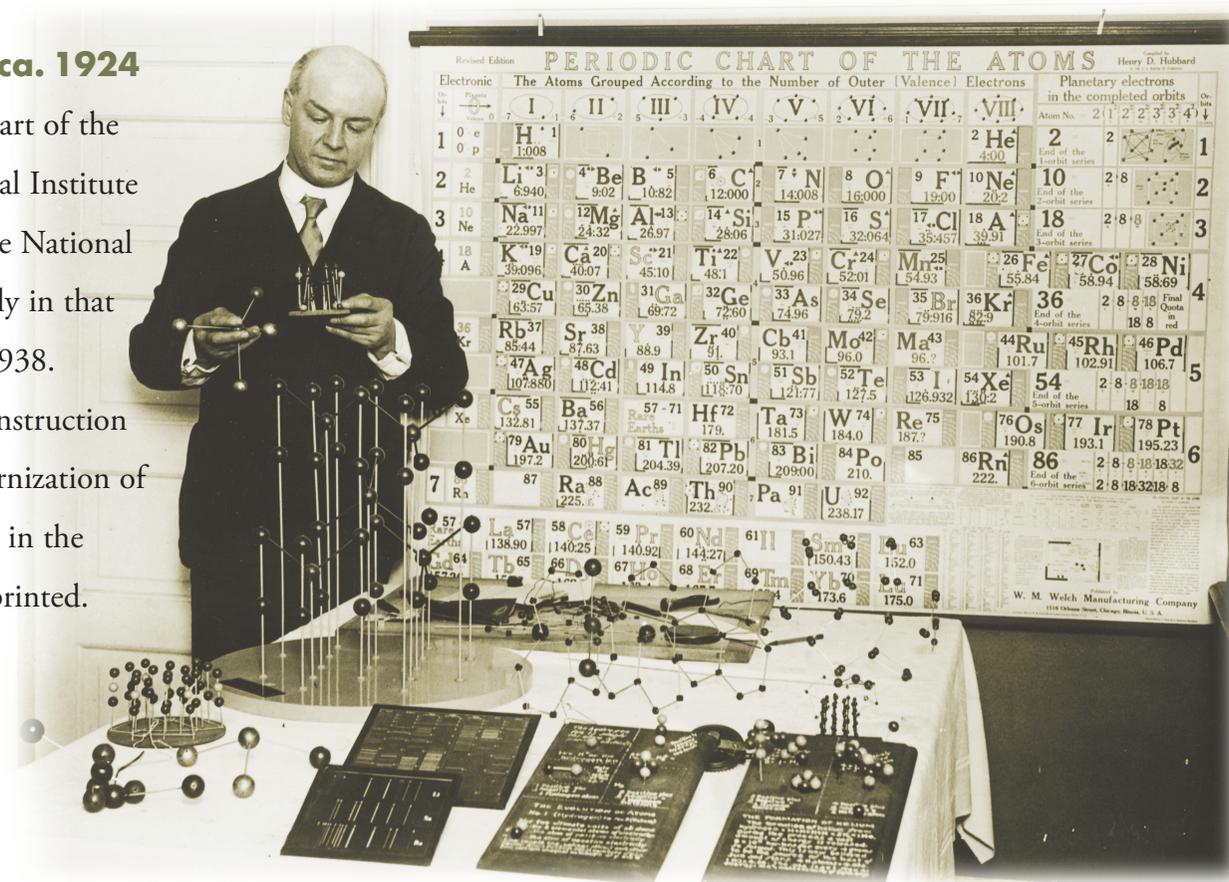
Group	PERIODIC TABLE																Physical Measurement Laboratory www.nist.gov/pml		Standard Reference Data www.nist.gov/srd										
1 IA											13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA													
1	<b>1</b> <sup>2</sup> S <sub>1/2</sub> <b>H</b> Hydrogen 1.008* 1s 13.5984											<b>5</b> <sup>2</sup> P <sub>1/2</sub> <b>B</b> Boron 10.81* 1s <sup>2</sup> 2s <sup>2</sup> 2p 8.2980	<b>6</b> <sup>3</sup> P <sub>0</sub> <b>C</b> Carbon 12.011* 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>2</sup> 11.2603	<b>7</b> <sup>4</sup> S <sub>3/2</sub> <b>N</b> Nitrogen 14.007* 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 14.5341	<b>8</b> <sup>3</sup> P <sub>2</sub> <b>O</b> Oxygen 15.999* 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>4</sup> 13.6181	<b>9</b> <sup>2</sup> P <sub>3/2</sub> <b>F</b> Fluorine 18.9984032 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>5</sup> 17.4228	<b>10</b> <sup>1</sup> S <sub>0</sub> <b>Ne</b> Neon 20.1797 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 21.5645												
2	<b>3</b> <sup>2</sup> S <sub>1/2</sub> <b>Li</b> Lithium 6.94* 1s <sup>2</sup> 2s 5.3917	<b>4</b> <sup>1</sup> S <sub>0</sub> <b>Be</b> Beryllium 9.012182 1s <sup>2</sup> 2s <sup>2</sup> 9.3227											<b>13</b> <sup>2</sup> P <sub>1/2</sub> <b>Al</b> Aluminum 26.9815386 [Ne]3s <sup>2</sup> 3p 5.9858	<b>14</b> <sup>3</sup> P <sub>0</sub> <b>Si</b> Silicon 28.085* [Ne]3s <sup>2</sup> 3p <sup>2</sup> 8.1517	<b>15</b> <sup>4</sup> S <sub>3/2</sub> <b>P</b> Phosphorus 30.973762 [Ne]3s <sup>2</sup> 3p <sup>3</sup> 10.4867	<b>16</b> <sup>3</sup> P <sub>2</sub> <b>S</b> Sulfur 32.06* [Ne]3s <sup>2</sup> 3p <sup>4</sup> 10.3600	<b>17</b> <sup>3</sup> P <sub>3/2</sub> <b>Cl</b> Chlorine 35.45* [Ne]3s <sup>2</sup> 3p <sup>5</sup> 12.9676	<b>18</b> <sup>1</sup> S <sub>0</sub> <b>Ar</b> Argon 39.948 [Ne]3s <sup>2</sup> 3p <sup>6</sup> 15.7596											
3	<b>11</b> <sup>2</sup> S <sub>1/2</sub> <b>Na</b> Sodium 22.98976928 [Ne]3s 5.1391	<b>12</b> <sup>1</sup> S <sub>0</sub> <b>Mg</b> Magnesium 24.3050 [Ne]3s <sup>2</sup> 7.6462	<b>3</b> IIIB	<b>4</b> IVB	<b>5</b> VB	<b>6</b> VIB	<b>7</b> VIIB	<b>8</b> VIII	<b>9</b> VIII	<b>10</b> VIII	<b>11</b> IB	<b>12</b> IIB																	
4	<b>19</b> <sup>2</sup> S <sub>1/2</sub> <b>K</b> Potassium 39.0983 [Ar]4s 4.3407	<b>20</b> <sup>1</sup> S <sub>0</sub> <b>Ca</b> Calcium 40.078 [Ar]4s <sup>2</sup> 6.1132	<b>21</b> <sup>2</sup> D <sub>3/2</sub> <b>Sc</b> Scandium 44.955912 [Ar]3d <sup>1</sup> 4s <sup>2</sup> 6.5615	<b>22</b> <sup>3</sup> F <sub>2</sub> <b>Ti</b> Titanium 47.867 [Ar]3d <sup>2</sup> 4s <sup>2</sup> 6.8281	<b>23</b> <sup>4</sup> F <sub>3/2</sub> <b>V</b> Vanadium 50.9415 [Ar]3d <sup>3</sup> 4s <sup>2</sup> 6.7462	<b>24</b> <sup>7</sup> S <sub>3</sub> <b>Cr</b> Chromium 51.9961 [Ar]3d <sup>5</sup> 4s 6.7665	<b>25</b> <sup>6</sup> S <sub>5/2</sub> <b>Mn</b> Manganese 54.938045 [Ar]3d <sup>5</sup> 4s <sup>2</sup> 7.4340	<b>26</b> <sup>5</sup> D <sub>4</sub> <b>Fe</b> Iron 55.845 [Ar]3d <sup>6</sup> 4s <sup>2</sup> 7.9025	<b>27</b> <sup>4</sup> F <sub>9/2</sub> <b>Co</b> Cobalt 58.933195 [Ar]3d <sup>7</sup> 4s <sup>2</sup> 7.8810	<b>28</b> <sup>3</sup> F <sub>4</sub> <b>Ni</b> Nickel 58.6934 [Ar]3d <sup>8</sup> 4s <sup>2</sup> 7.6399	<b>29</b> <sup>2</sup> S <sub>1/2</sub> <b>Cu</b> Copper 63.546 [Ar]3d <sup>10</sup> 4s 9.3942	<b>30</b> <sup>1</sup> S <sub>0</sub> <b>Zn</b> Zinc 65.38 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 9.3942	<b>31</b> <sup>2</sup> P <sub>1/2</sub> <b>Ga</b> Gallium 69.723 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p 5.9993	<b>32</b> <sup>3</sup> P <sub>0</sub> <b>Ge</b> Germanium 72.63 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>2</sup> 7.8994	<b>33</b> <sup>4</sup> S <sub>3/2</sub> <b>As</b> Arsenic 74.92160 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>3</sup> 9.7886	<b>34</b> <sup>3</sup> P <sub>2</sub> <b>Se</b> Selenium 78.96 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 9.7524	<b>35</b> <sup>2</sup> P <sub>3/2</sub> <b>Br</b> Bromine 79.904 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 11.8138	<b>36</b> <sup>1</sup> S <sub>0</sub> <b>Kr</b> Krypton 83.798 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>6</sup> 13.9996											
5	<b>37</b> <sup>2</sup> S <sub>1/2</sub> <b>Rb</b> Rubidium 85.4678 [Kr]5s 4.1771	<b>38</b> <sup>1</sup> S <sub>0</sub> <b>Sr</b> Strontium 87.62 [Kr]5s <sup>2</sup> 5.6949	<b>39</b> <sup>2</sup> D <sub>3/2</sub> <b>Y</b> Yttrium 88.90585 [Kr]4d <sup>5</sup> 5s <sup>2</sup> 6.2173	<b>40</b> <sup>3</sup> F <sub>2</sub> <b>Zr</b> Zirconium 91.224 [Kr]4d <sup>2</sup> 5s <sup>2</sup> 6.6339	<b>41</b> <sup>6</sup> D <sub>1/2</sub> <b>Nb</b> Niobium 92.90638 [Kr]4d <sup>4</sup> 5s 6.7589	<b>42</b> <sup>7</sup> S <sub>3</sub> <b>Mo</b> Molybdenum 95.96 [Kr]4d <sup>5</sup> 5s 7.0924	<b>43</b> <sup>6</sup> S <sub>5/2</sub> <b>Tc</b> Technetium (98) [Kr]4d <sup>5</sup> 5s <sup>2</sup> 7.1194	<b>44</b> <sup>5</sup> F <sub>5</sub> <b>Ru</b> Ruthenium 101.07 [Kr]4d <sup>7</sup> 5s 7.3605	<b>45</b> <sup>4</sup> F <sub>9/2</sub> <b>Rh</b> Rhodium 102.90550 [Kr]4d <sup>8</sup> 5s 7.4589	<b>46</b> <sup>1</sup> S <sub>0</sub> <b>Pd</b> Palladium 106.42 [Kr]4d <sup>10</sup> 8.3369	<b>47</b> <sup>2</sup> S <sub>1/2</sub> <b>Ag</b> Silver 107.8682 [Kr]4d <sup>10</sup> 5s 7.5762	<b>48</b> <sup>1</sup> S <sub>0</sub> <b>Cd</b> Cadmium 112.411 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 8.9938	<b>49</b> <sup>2</sup> P <sub>1/2</sub> <b>In</b> Indium 114.818 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p 7.5864	<b>50</b> <sup>3</sup> P <sub>0</sub> <b>Sn</b> Tin 118.710 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>2</sup> 7.3439	<b>51</b> <sup>4</sup> S <sub>3/2</sub> <b>Sb</b> Antimony 121.760 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>3</sup> 8.6084	<b>52</b> <sup>3</sup> P <sub>2</sub> <b>Te</b> Tellurium 127.60 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>4</sup> 9.0097	<b>53</b> <sup>2</sup> P <sub>3/2</sub> <b>I</b> Iodine 126.90447 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>5</sup> 10.4513	<b>54</b> <sup>1</sup> S <sub>0</sub> <b>Xe</b> Xenon 131.293 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>6</sup> 12.1298											
6	<b>55</b> <sup>2</sup> S <sub>1/2</sub> <b>Cs</b> Cesium 132.9054519 [Xe]6s 3.8939	<b>56</b> <sup>1</sup> S <sub>0</sub> <b>Ba</b> Barium 137.327 [Xe]6s <sup>2</sup> 5.2117	<div style="display: flex; justify-content: space-between;"> <div style="writing-mode: vertical-rl; transform: rotate(180deg);">Lanthanides</div> <div style="writing-mode: vertical-rl; transform: rotate(180deg);">Actinides</div> </div>										<b>72</b> <sup>3</sup> F <sub>2</sub> <b>Hf</b> Hafnium 178.49 [Xe]4f <sup>14</sup> 5d <sup>2</sup> 6s <sup>2</sup> 6.8251	<b>73</b> <sup>4</sup> F <sub>3/2</sub> <b>Ta</b> Tantalum 180.94788 [Xe]4f <sup>14</sup> 5d <sup>3</sup> 6s <sup>2</sup> 7.5496	<b>74</b> <sup>5</sup> D <sub>0</sub> <b>W</b> Tungsten 183.84 [Xe]4f <sup>14</sup> 5d <sup>4</sup> 6s <sup>2</sup> 7.8640	<b>75</b> <sup>6</sup> S <sub>5/2</sub> <b>Re</b> Rhenium 186.207 [Xe]4f <sup>14</sup> 5d <sup>5</sup> 6s <sup>2</sup> 7.8335	<b>76</b> <sup>5</sup> D <sub>4</sub> <b>Os</b> Osmium 190.23 [Xe]4f <sup>14</sup> 5d <sup>6</sup> 6s <sup>2</sup> 8.4382	<b>77</b> <sup>4</sup> F <sub>9/2</sub> <b>Ir</b> Iridium 192.217 [Xe]4f <sup>14</sup> 5d <sup>7</sup> 6s <sup>2</sup> 8.9670	<b>78</b> <sup>3</sup> D <sub>3</sub> <b>Pt</b> Platinum 195.084 [Xe]4f <sup>14</sup> 5d <sup>8</sup> 6s <sup>2</sup> 8.9588	<b>79</b> <sup>2</sup> S <sub>1/2</sub> <b>Au</b> Gold 196.966569 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s 9.2256	<b>80</b> <sup>1</sup> S <sub>0</sub> <b>Hg</b> Mercury 200.59 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 10.4375	<b>81</b> <sup>2</sup> P <sub>1/2</sub> <b>Tl</b> Thallium 204.38* [Hg]6p 6.1083	<b>82</b> <sup>3</sup> P <sub>0</sub> <b>Pb</b> Lead 207.2 [Hg]6p <sup>2</sup> 7.4167	<b>83</b> <sup>4</sup> S <sub>3/2</sub> <b>Bi</b> Bismuth 208.98040 [Hg]6p <sup>3</sup> 7.2855	<b>84</b> <sup>3</sup> P <sub>2</sub> <b>Po</b> Polonium (209) [Hg]6p <sup>4</sup> 8.414	<b>85</b> <sup>2</sup> P <sub>3/2</sub> <b>At</b> Astatine (210) [Hg]6p <sup>5</sup> 9.350	<b>86</b> <sup>1</sup> S <sub>0</sub> <b>Rn</b> Radon (222) [Hg]6p <sup>6</sup> 10.7485		
7	<b>87</b> <sup>2</sup> S <sub>1/2</sub> <b>Fr</b> Francium (223) [Rn]7s 4.0727	<b>88</b> <sup>1</sup> S <sub>0</sub> <b>Ra</b> Radium (226) [Rn]7s <sup>2</sup> 5.2784											<b>104</b> <sup>3</sup> F <sub>2</sub> <b>Rf</b> Rutherfordium (265) [Rn]5f <sup>14</sup> 6d <sup>2</sup> 7s <sup>2</sup> 6.01	<b>105</b> <sup>4</sup> F <sub>3/2</sub> <b>Db</b> Dubnium (268) [Rn]5f <sup>14</sup> 6d <sup>3</sup> 7s <sup>2</sup> 6.8	<b>106</b> <sup>5</sup> D <sub>0</sub> <b>Sg</b> Seaborgium (271) [Rn]5f <sup>14</sup> 6d <sup>4</sup> 7s <sup>2</sup> 7.8	<b>107</b> <sup>6</sup> S <sub>5/2</sub> <b>Bh</b> Bohrium (270) [Rn]5f <sup>14</sup> 6d <sup>5</sup> 7s <sup>2</sup> 7.7	<b>108</b> <sup>5</sup> D <sub>4</sub> <b>Hs</b> Hassium (277) [Rn]5f <sup>14</sup> 6d <sup>6</sup> 7s <sup>2</sup> 7.6	<b>109</b> <sup>4</sup> F <sub>9/2</sub> <b>Mt</b> Meitnerium (276) [Rn]5f <sup>14</sup> 6d <sup>7</sup> 7s <sup>2</sup> 7.6	<b>110</b> <sup>3</sup> D <sub>3</sub> <b>Ds</b> Darmstadtium (281) [Rn]5f <sup>14</sup> 6d <sup>8</sup> 7s <sup>2</sup> 7.6	<b>111</b> <sup>2</sup> S <sub>1/2</sub> <b>Rg</b> Roentgenium (280) [Rn]5f <sup>14</sup> 6d <sup>9</sup> 7s <sup>2</sup> 7.6	<b>112</b> <sup>1</sup> S <sub>0</sub> <b>Cn</b> Copernicium (285) [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7.6	<b>113</b> <b>Uut</b> Ununtrium (284) [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>1</sup> 7.6	<b>114</b> <b>Fl</b> Flerovium (289) [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>2</sup> 7.6	<b>115</b> <b>Uup</b> Ununpentium (288) [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>3</sup> 7.6	<b>116</b> <b>Lv</b> Livermorium (293) [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>4</sup> 7.6	<b>117</b> <b>Uus</b> Ununseptium (294) [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>5</sup> 7.6	<b>118</b> <b>Uuo</b> Ununoctium (294) [Rn]5f <sup>14</sup> 6d <sup>10</sup> 7s <sup>2</sup> 7p <sup>6</sup> 7.6		
		<b>57</b> <sup>2</sup> D <sub>3/2</sub> <b>La</b> Lanthanum 138.90547 [Xe]5d <sup>1</sup> 6s <sup>2</sup> 5.5769	<b>58</b> <sup>1</sup> G <sub>4</sub> <b>Ce</b> Cerium 140.116 [Xe]4f <sup>1</sup> 5d <sup>1</sup> 6s <sup>2</sup> 5.5386	<b>59</b> <sup>4</sup> I <sub>9/2</sub> <b>Pr</b> Praseodymium 140.90765 [Xe]4f <sup>3</sup> 6s <sup>2</sup> 5.473	<b>60</b> <sup>5</sup> I <sub>7/2</sub> <b>Nd</b> Neodymium 144.242 [Xe]4f <sup>4</sup> 6s <sup>2</sup> 5.5250	<b>61</b> <sup>6</sup> H <sub>9/2</sub> <b>Pm</b> Promethium (145) [Xe]4f <sup>5</sup> 6s <sup>2</sup> 5.582	<b>62</b> <sup>7</sup> F <sub>0</sub> <b>Sm</b> Samarium 150.36 [Xe]4f <sup>6</sup> 6s <sup>2</sup> 5.6437	<b>63</b> <sup>8</sup> G <sub>7/2</sub> <b>Eu</b> Europium 151.964 [Xe]4f <sup>7</sup> 6s <sup>2</sup> 5.6704	<b>64</b> <sup>9</sup> D <sub>2</sub> <b>Gd</b> Gadolinium 157.25 [Xe]4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup> 6.1498	<b>65</b> <sup>6</sup> H <sub>15/2</sub> <b>Tb</b> Terbium 158.92535 [Xe]4f <sup>9</sup> 6s <sup>2</sup> 5.9391	<b>66</b> <sup>5</sup> F <sub>8</sub> <b>Dy</b> Dysprosium 162.500 [Xe]4f <sup>10</sup> 6s <sup>2</sup> 6.0215	<b>67</b> <sup>4</sup> I <sub>15/2</sub> <b>Ho</b> Holmium 164.93032 [Xe]4f <sup>11</sup> 6s <sup>2</sup> 6.0215	<b>68</b> <sup>3</sup> H <sub>6</sub> <b>Er</b> Erbium 167.259 [Xe]4f <sup>12</sup> 6s <sup>2</sup> 6.1077	<b>69</b> <sup>2</sup> F <sub>7/2</sub> <b>Tm</b> Thulium 168.93421 [Xe]4f <sup>13</sup> 6s <sup>2</sup> 6.1843	<b>70</b> <sup>1</sup> S <sub>0</sub> <b>Yb</b> Ytterbium 173.054 [Xe]4f <sup>14</sup> 6s <sup>2</sup> 6.2542	<b>71</b> <sup>2</sup> D <sub>3/2</sub> <b>Lu</b> Lutetium 174.9668 [Xe]4f <sup>14</sup> 5d <sup>1</sup> 6s <sup>2</sup> 5.4259													
		<b>89</b> <sup>2</sup> D <sub>3/2</sub> <b>Ac</b> Actinium (227) [Rn]6d <sup>1</sup> 7s <sup>2</sup> 5.3802	<b>90</b> <sup>3</sup> F <sub>2</sub> <b>Th</b> Thorium 232.03806 [Rn]6d <sup>2</sup> 7s <sup>2</sup> 6.3067	<b>91</b> <sup>4</sup> K <sub>11/2</sub> <b>Pa</b> Protactinium 231.03588 [Rn]5f <sup>1</sup> 6d <sup>1</sup> 7s <sup>2</sup> 5.89	<b>92</b> <sup>5</sup> L <sub>9</sub> <b>U</b> Uranium 238.02891 [Rn]5f <sup>3</sup> 6d <sup>1</sup> 7s <sup>2</sup> 6.1941	<b>93</b> <sup>6</sup> L <sub>11/2</sub> <b>Np</b> Neptunium (237) [Rn]5f <sup>4</sup> 6d <sup>1</sup> 7s <sup>2</sup> 6.2655	<b>94</b> <sup>7</sup> F <sub>0</sub> <b>Pu</b> Plutonium (244) [Rn]5f <sup>6</sup> 7s <sup>2</sup> 6.0258	<b>95</b> <sup>8</sup> G <sub>7/2</sub> <b>Am</b> Americium (243) [Rn]5f <sup>7</sup> 7s <sup>2</sup> 5.9738	<b>96</b> <sup>9</sup> D <sub>2</sub> <b>Cm</b> Curium (247) [Rn]5f <sup>8</sup> 6d <sup>1</sup> 7s <sup>2</sup> 5.9914	<b>97</b> <sup>6</sup> H <sub>15/2</sub> <b>Bk</b> Berkelium (247) [Rn]5f <sup>9</sup> 7s <sup>2</sup> 6.1978	<b>98</b> <sup>5</sup> F <sub>8</sub> <b>Cf</b> Californium (251) [Rn]5f <sup>10</sup> 7s <sup>2</sup> 6.2817	<b>99</b> <sup>4</sup> I <sub>15/2</sub> <b>Es</b> Einsteinium (252) [Rn]5f <sup>11</sup> 7s <sup>2</sup> 6.3676	<b>100</b> <sup>3</sup> H <sub>6</sub> <b>Fm</b> Fermium (257) [Rn]5f <sup>12</sup> 7s <sup>2</sup> 6.50	<b>101</b> <sup>2</sup> F <sub>7/2</sub> <b>Md</b> Mendelevium (258) [Rn]5f <sup>13</sup> 7s <sup>2</sup> 6.58	<b>102</b> <sup>1</sup> S <sub>0</sub> <b>No</b> Nobelium (259) [Rn]5f <sup>14</sup> 7s <sup>2</sup> 6.65	<b>103</b> <sup>2</sup> P <sub>3/2</sub> <b>Lr</b> Lawrencium (262) [Rn]5f <sup>14</sup> 7d <sup>1</sup> 7p <sup>0</sup> <													

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## The Hubbard Chart of the Atoms, ca. 1924

Henry D. Hubbard, the designer of the “Chart of the Atoms,” was the first secretary of the National Institute of Standards and Technology (then-called the National Bureau of Standards) and served continuously in that capacity from 1901 until his retirement in 1938.

Secretary Hubbard made a contribution to instruction in physics that is still in use today, his modernization of Mendeleev’s periodic table. First constructed in the 1920s, it has been frequently revised and reprinted.



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